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(54) **REFRIGERATOR OIL COMPOSITION FOR HYDROFLUOROCARBON REFRIGERANT.**

(57) A refrigerator oil composition for a hydrofluorocarbon (hydrogen-containing flon) refrigerant, characterized by comprising a polyoxyalkylene glycol derivative and/or a specified polyester compound and added thereto (a) a partial fatty acid ester of a polyhydric alcohol and (b) a phosphate compound and/or a phosphite compound. It is excellent in the compatibility with a hydrogen-containing flon refrigerant, such as flon (134a), and in the lubricity, and utilizable as a refrigerator oil useful for an improvement in the resistance to abrasion, particularly between an aluminum material and a steel material.

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TECHNICAL FIELD

The present invention relates to a refrigerator oil composition for hydrogen-containing hydrofluorocarbon refrigerant. More particularly, it relates to a refrigerator oil composition which has favorable miscibility with a hydrogen-containing hydrofluorocarbon (hydrogenated fluoroalkane) (hereinafter referred to as "hydrogenated Flon compound") such as 1,1,1,2-tetrafluoroethane (hereinafter referred to as "Flon 134a") capable of being used as a replacement of a chlorofluorocarbon (Flon compound) such as dichlorodifluoromethane (hereinafter referred to as "Flon 12") as a refrigerant now causing a serious environmental pollution problem, and at the same time, which is excellent in lubrication performance as well as wear resistance.

BACKGROUND ART

Polyoxyalkylene glycol derivatives have recently been reported as being effective as a lubricating oil for a refrigerator employing a hydrogenated Flon refrigerant such as Flon 134a without a fear of causing environmental pollution problem by virtue of their high miscibility with the above-mentioned hydrogenated Flon refrigerant (Specification of US PAT. No. 4,755,316).

Nevertheless, the polyoxyalkylene glycol derivatives are poor in wear resistance and increase the wear between aluminum material and steel material in a refrigerator in an atmosphere of aforementioned refrigerant, thus causing serious problem in practical application. The frictional surfaces between aluminum material and steel material include the contact surface of a piston and piston shoe for a reciprocating compressor, and that of a vane and housing for a rotary compressor. In either case, lubrication is important for the friction surfaces.

On the other hand, the conventional extreme pressure agents which have heretofore been used in refrigerator oil impaired the stability of the oil and suffered insufficient wear-resistance effect, making themselves impossible to endure practical application.

DISCLOSURE OF INVENTION

In order to develop a refrigerator oil (lubricating oil) which is excellent in lubrication performance as well as miscibility with hydrogenated Flon refrigerant such as Flon 134a, and at the same time, effective for improving wear resistance, especially the wear resistance between aluminum material and steel material, intensive research and investigation have been made by the present inventors. As a result, it has been found that the above-mentioned purpose can be attained by compounding (a) an aliphatic acid partially esterified with a polyhydric alcohol and (b) a phosphate compound and/or phosphite compound in a polyoxyalkylene glycol and/or a specific polyester compound. The present invention has been accomplished on the basis of such a finding.

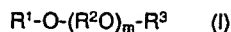
That is, the present invention provides a refrigerator oil composition for hydrogen-containing hydrofluorocarbon refrigerant which comprises at least one compound selected from (A) a polyoxyalkylene glycol derivative, and (B) polyester compounds having a kinematic viscosity at 40 °C of 5 to 1000 cSt and at least two ester linkages which are compounded with (a) an aliphatic acid partially esterified with a polyhydric alcohol and (b) at least one compound selected from phosphate compounds and phosphite compounds.

Furthermore, the present invention provides a method for effecting lubrication in a compression-type refrigerator using a hydrofluorocarbon as a refrigerant characterized in that the lubrication is effected by the use of the above-mentioned refrigerator oil composition, and still furthermore, the present invention provides a compression-type refrigeration system which comprises compressor(s), the hydrogen-containing hydrofluorocarbon as a refrigerant and the above-mentioned refrigerator oil composition as a lubricating oil.

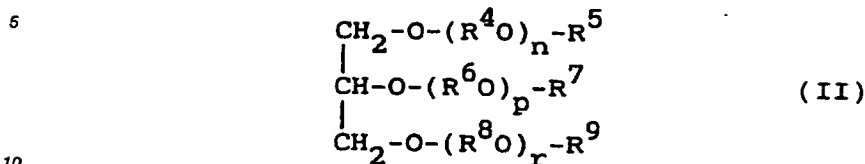
BEST MODE FOR CARRYING OUT THE INVENTION

The refrigerator oil composition according to the present invention comprises, as essential components, (A) a polyoxyalkylene glycol derivative, (B) a polyester compound having a kinematic viscosity at 40 °C of 5 to 1000 cSt and at least two ester linkages, or a mixture of the above (A) and (B).

The polyoxyalkylene glycol derivative as the component (A) includes a variety of compounds, which may be optionally selected according to the purpose of use. A suitable polyoxyalkylene glycol derivative is exemplified by a compound represented by the formula



and/or



wherein R^1 , R^3 , R^5 , R^7 and R^9 are each a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an alkenyl group having 2 to 20 carbon atoms, an alkynyl group having 2 to 20 carbon atoms, an acyl group having 1 to 20 carbon atoms, an aryl group having 6 to 20 carbon atoms, an aralkyl group having 7 to 20 carbon atoms, an alkylene group having 2 to 4 carbon atoms, an alkylene group having 2 to 4 carbon atoms; R^2 , R^4 , R^6 and R^8 are each an alkylene group having 2 to 4 carbon atoms; m is an integer from 3 through 100; and n , p and r are each a positive integer simultaneously satisfying the relationship

$$3 \leq n + p + r \leq 100.$$

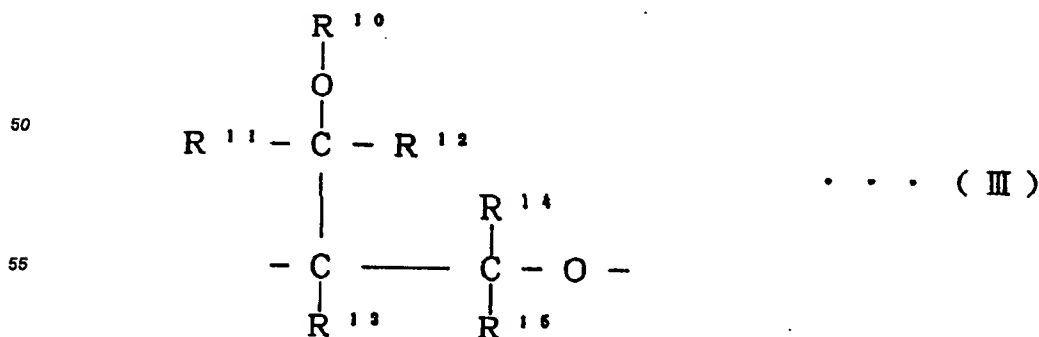
The foregoing compound may be used alone or in combination with each other. As mentioned before, R^1 , R^3 , R^5 , R^7 and R^9 are each, in addition to a hydrogen atom; an alkyl group having 1 to 20 carbon atoms such as methyl, ethyl, propyl, butyl; an alkenyl group having 2 to 20 carbon atoms such as vinyl, allyl, butenyl; an alkynyl group having 2 to 20 carbon atoms such as ethynyl, propynyl; an aryl group having 6 to 20 carbon atoms such as phenyl, tolyl, xylol, nonylphenyl, dodecylphenyl; an aralkyl group having 7 to 20 carbon atoms such as benzyl, phenetyl; or a cycloalkyl group having 4 to 20 carbon atoms such as cyclohexyl, cycloheptyl, methylcyclohexyl, nonylcyclohexyl. Among them, a hydrogen atom and an alkyl group having 1 to 10 carbon atoms are preferable. In particular, a compound of the formula (I) wherein R^1 and R^3 are simultaneously methyl is most suitable.

As mentioned before, R^2 , R^4 , R^6 and R^8 are each an alkylene group having 2 to 4 carbon atoms such as a group of ethylene, propylene or butylene. In the formula (I), m is an integer from 3 through 100, preferably 6 through 60, and n , p and r in the formula (II) are each a positive integer simultaneously satisfying the relationship $3 \leq n + p + r \leq 100$, preferably $6 \leq n + p + r \leq 60$.

The compound represented by the general formula (I) or (II) has a kinematic viscosity at 100°C of preferably 2 to 250 cSt. The compound represented by the general formula (I) has a kinematic viscosity at 100°C of desirably 5 to 50 cSt, more desirably 6 (when $m=12$) to 50 cSt, still more desirably 7 (when $m=14$) to 50 cSt, most desirably 9 (when $m=19$) to 50 cSt.

In the formula (I) & (II), it is preferable that R^2O , R^4O , R^6O and R^8O are each a copolymer containing ethyleneoxide unit of 30% by weight or less, for example, a copolymer of ethylene oxide unit and propylene oxide unit.

The polyoxyalkylene glycol derivative to be used in the present invention may be exemplified, in addition to the above-mentioned compounds, by a compound having at least one constitutional unit represented by the general formula



wherein R^{10} indicates a monovalent hydrocarbon radical having 1 to 10 carbon atoms, and R^{11} through R^{15} stand for a hydrogen atom or a monovalent hydrocarbon radical having 1 to 10 carbon atoms.

In the refrigerator oil composition of the present invention, there may be used, as an essential component, a polyester compound having a kinematic viscosity at 40°C of 5 to 1000 cSt, preferably 10 to 500 cSt and at least two ester linkages in place of (A) a polyoxyalkylene glycol derivative. A variety of compounds are available as the aforementioned polyester compound and may be optionally selected according to the purpose of use. Among them, a suitable compound is exemplified by reaction products represented by the following (IV) to (VIII).

(IV) a reaction product of (1) a polybasic carboxylic acid or derivative thereof, (2) a polyhydric alcohol or derivative thereof and (3) a monobasic aliphatic acid or derivative thereof.

(V) a reaction product of (1) a polybasic carboxylic acid or derivative thereof, (2) a polyhydric alcohol or derivative thereof and (4) a monohydric aliphatic alcohol or derivative thereof.

(VI) a reaction product of (2) a polyhydric alcohol or derivative thereof and (3) a monobasic aliphatic acid or derivative thereof, preferably a equivalent reaction product.

(VII) a reaction product of (4) a monohydric aliphatic alcohol or derivative thereof and (1) a polybasic carboxylic acid or derivative thereof.

(VIII) a reaction product of (1) a polybasic carboxylic acid or derivative thereof and (2) a polyhydric alcohol or derivative thereof.

Among a variety of (1) polybasic carboxylic acids, preferable ones include an aliphatic saturated dicarboxylic acid having 2 to 12 carbon atoms such as oxalic acid, malonic acid, succinic acid, glutaric acid, adipic acid, pimelic acid, suberic acid, azelaic acid, sebacic acid, dodecanedioic acid; an aliphatic unsaturated dicarboxylic acid having 4 to 14 carbon atoms such as maleic acid, fumaric acid, alkenyl succinic acid; an aromatic dicarboxylic acid having 8 carbon atoms such as phthalic acid, isophthalic acid, terephthalic acid; a dicarboxylic acid such as epoxide including epoxyhexahydrophthalic acid, etc.; and polybasic (tribasic or more) such as citric acid, trimellitic acid, pyromellitic acid.

Derivatives of polybasic carboxylic acid are exemplified by monoester, diester, metal salt, anhydride, acid chloride and the like each of polybasic carboxylic acid.

Examples of (2) polyhydric alcohol include glycols such as ethylene glycol, diethylene glycol, triethylene glycol, tetraethylene glycol, polyethylene glycol, propylene glycol, dipropylene glycol, polypropylene glycol, 1,2-butylene glycol, 1,3-butylene glycol, 1,4-butylene glycol, 2,3-butylene glycol, polybutylene glycol, 2-methyl-2,4-pentanediol, 2-ethyl-1,3-hexanediol; glycerol; hindered alcohol such as neopentyl glycol, trimethylol propane, pentaerythritol, dipentaerythritol; sorbitol; sorbitan and the like. Derivatives of polyhydric alcohol are exemplified by chloride, metal salt thereof.

Among a variety of (3) monobasic aliphatic acids (primary, secondary, and tertiary), those having an alkyl group having 1 to 20 carbon atoms, particularly branched-chain alkyl group having 3 to 18 carbon atoms, more particularly 4 to 12 carbon atoms are preferable because of their favorable miscibility with Flon as a refrigerant, that is, the higher miscibility temperature at elevated temperatures. Specific examples of such monobasic aliphatic acids include acetic acid, propionic acid, butyric acid, isobutyric acid, pivalic acid, n-valeric acid, isovaleric acid, caproic acid, 2-ethylbutyric acid, n-caproic acid, 2-methylcaproic acid, n-heptylic acid, n-octanoic acid, 2-ethylhexanoic acid, 3,5,5-trimethylhexanoic acid, nonanoic acid, tert-nonanoic acid, dodecanoic acid, tert-dodecanoic acid, lauric acid and the like. Derivatives of such monobasic aliphatic acids are exemplified by ester, metal salt, acid chloride, anhydride thereof and the like.

Among a variety of (4) monohydric aliphatic alcohols or derivatives thereof, those having an alkyl group having 1 to 20 carbon atoms, particularly branched-chain alkyl group having 3 to 18 carbon atoms, more particularly 4 to 12 carbon atoms are preferable because of their favorable miscibility with Flon as a refrigerant, that is, the higher miscibility temperature at elevated temperatures. Specific examples of such monohydric aliphatic alcohols include methyl alcohol, ethyl alcohol, n-propyl alcohol, isopropyl alcohol, n-butyl alcohol, isobutyl alcohol, sec-butyl alcohol, tert-butyl alcohol, n-amyl alcohol, isoamyl alcohol, tert-amyl alcohol, diethyl carbinol, n-hexyl alcohol, methylamyl alcohol, ethylbutyl alcohol, heptyl alcohol, methylamyl carbinol, dimethylpentyl alcohol, n-octyl alcohol, sec-octyl alcohol, 2-ethylhexyl alcohol, isooctyl alcohol, n-nonyl alcohol, diisobutyl carbinol, n-decyl alcohol, isodecyl alcohol and the like. Derivatives of such monohydric aliphatic alcohols are exemplified by chloride, metal salt thereof and the like.

It is desirable that a polyester compound to be used in the present invention has a specific range of kinematic viscosity and at the same time, is composed of at least one of the reaction products as defined by the above-described (IV) to (VIII). A reaction product (IV) is obtained by reacting the above compounds (1), (2) and (3) with each other. Though the chemical structures of the reaction products have not yet been clarified, each of two carboxyl groups (or groups derived from the carboxyl groups in the case of derivative) of a (1) polybasic carboxylic acid usually reacts with one of hydroxy groups (or groups derived from the

hydroxy groups in the case of derivative) of a (2) polyhydric alcohol to combine together and further, another hydroxy group of the (2) polyhydric alcohol usually reacts with the carboxyl group of a (3) monobasic aliphatic acid to combine together. The residual carboxyl groups of the (1) polybasic carboxylic acid and the residual hydroxyl groups of the (2) polyhydric alcohol may remain unreacted or may react with

A reaction product (V) is obtained by reacting the above compounds (1) (2), and (4) with each other. Although the chemical structures of the reaction products have not yet been clarified, each of two hydroxy groups of a (2) polyhydric alcohol usually reacts with a carboxyl group of a (1) polybasic carboxylic acid to combine together and further, the residual carboxyl group of the (1) polybasic acid usually reacts with the hydroxy group of a (4) monohydric aliphatic alcohol to combine together. The residual carboxyl groups of the polybasic carboxylic acid and the residual hydroxyl groups of the (2) polyhydric alcohol may remain unreacted or may react with other functional groups.

The use of the mixture of the component (A) and component (B) as the essential component of the refrigerator oil composition (that is, the base oil of lubricating oil) of the present invention is also effective. In this case, the mixing ratio by weight of the component (A) to the component (B) is not particularly limited, but desirably 5 to 95/95 to 5, particularly desirably 10 to 90/90 to 10.

The use of a mineral oil or a synthetic oil other than the above-mentioned (A) polyoxyalkylene glycol derivative or (B) polyester compound as the essential component of the refrigerator oil composition (that is, the base oil of lubricating oil) of the present invention causes insufficient miscibility with a hydrogenated

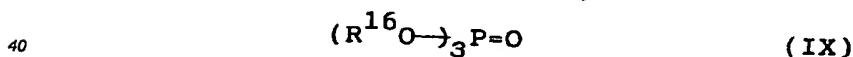
Flon refrigerant such as Flon 134a.

It is necessary to compound (a) an aliphatic acid partially esterified with a polyhydric alcohol and (b) a phosphate compound and/or phosphite compound with the aforementioned (A) a polyoxyalkylene glycol derivative and/or (B) a polyester compound.

By the term "an aliphatic acid partially, esterified with a polyhydric alcohol" as used herein is meant a partially esterified product formed from a polyhydric alcohol such as glycol, glycerol, trimethylol propane, pentaerythritol, sorbitan, sorbitol or the like and a saturated or unsaturated straight-chain or branched-chain aliphatic acid having 1 to 24 carbon atoms, preferably from glycerol, sorbitan or sorbitol and an aliphatic acid having 8 to 22 carbon atoms. In particular, an monobasic aliphatic acid ester is most suitable.

In the refrigerator oil composition according to the present invention, the compounding ratio of (a) an aliphatic acid partially esterified with a polyhydric alcohol is not particularly limited and may be optionally selected depending upon the situation, but desirably 0.1 to 10%, particularly desirably 0.5 to 5% each by weight based on the total amount of the composition.

As a component (b), either a phosphate compound or a phosphite compound or both of them is used. In this case, phosphate compounds may be roughly classified into alkyl phosphate compounds and aryl phosphate compounds. A suitable phosphate compound can be exemplified by the compound represented by the general formula



wherein R^{16} is a hydrocarbon radical or chlorinated hydrocarbon radical having a total number of 15 or more carbon atoms, particularly an alkyl group having 8 to 20 carbon atoms (straight-chain or branched-chain, saturated or unsaturated), phenyl group, alkyl-substituted phenyl group having 1 to 12 carbon atoms, chlorinated phenyl group, or chlorinated alkylphenyl group, and may be the same or different. Specific examples include tricresyl phosphate (TCP), triphenyl phosphate, triisopropyl-phenyl phosphate, trioctyl phosphate, trilauryl phosphate, tristearyl phosphate, trioleyl phosphate, diphenyloctyl phosphate, o-, m-, p-monochlorophenyl phosphate, dichlorophenyl phosphate, monochlorotolyl phosphate, dichlorotolyl phosphate and the like. The use of tricresyl phosphate is particularly desirable.

The phosphite compounds to be used as component (b) include a variety of compounds and may be roughly classified into alkyl phosphite compounds and aryl phosphite compounds. A suitable phosphite compound can be exemplified by the compound represented by the general formula



wherein R^{17} is a hydrogen atom or a hydrocarbon radical having a total number of 15 or more carbon

atoms, particularly an alkyl group having 8 to 20 carbon atoms (straight-chain or branched-chain, saturated or unsaturated), phenyl group, alkyl-substituted phenyl group having 1 to 12 carbon atoms, and may be the same or different, but excluding the case where two or more of three R¹⁷s are each a hydrogen atom. Specific examples include trioctyl phosphite, triauryl phosphite, tristearyl phosphite, trioctyl phosphate, triphenyl phosphite, tricresyl phosphite, tris(nonylphenyl) phosphite, diphenyldecyl phosphite, dioctyl hydrogen phosphite, dilauryl hydrogen phosphite, dioleil hydrogen phosphite, di(nonylphenyl) hydrogen phosphite and the like.

As a component (b) in the present invention, either a phosphate compound or a phosphite compound or both of them is used. The compounding ratio of component (b) is not particularly limited, but is optionally selected in the range of 0.1 to 5% by weight, preferably 0.3 to 3% by weight based on the total amount of composition.

The composition of the present invention comprises as an essential component (A) a polyoxyalkylene glycol derivative and/or (B) a specific polyester compound, said essential component being compounded with (a) an aliphatic acid partially esterified with a polyhydric alcohol and (b) a phosphate compound and/or phosphite compound. Moreover, various additives that are used in the conventional lubricating oils such as load carrying additives (extreme pressure agent, oiliness agent, anti-wear additive, etc.), chlorine capturing agent, antioxidants, metal deactivators, defoaming agents, detergent-dispersants, viscosity index improvers, antirust agents, corrosion inhibitors, pour point depressants, etc. may be optionally blended when necessary.

In addition, mineral oils or synthetic oils that are generally used as base oil of lubricating oil may be used together with the above-mentioned (A) a polyoxyalkylene glycol derivative or (B) a polyester compound according to the demand.

As the load carrying additives, there can be mentioned organic sulfide additives such as monosulfides, polysulfides, sulfoxides, sulfones, thiosulfates, sulfurized oil and fat, thiocarbonates, thiophenes, thiazoles, methanesulfonic acid esters, etc.; phosphate additives such as phosphoric monoesters, phosphoric diesters, phosphoric triesters (excluding the ester of the general formula (IX)) etc.; phosphite additives (excluding the ester of the general formula (X)) such as phosphorus monoesters, phosphorus diesters, phosphorus triesters, etc.; thiophosphate additives such as thiophosphoric acid triesters; fatty acid additives such as higher fatty acids, hydroxyaryl fatty acids, carboxylic acid-containing polyhydric alcohol esters, metallic soap, etc.; fatty acid ester additives such as polyhydric alcohol esters, acrylic esters, etc.; organic chlorine additives such as chlorinated hydrocarbons, chlorinated carboxylic acid derivatives, etc.; organic fluorine additive such as fluorinated aliphatic carboxylic acids, fluoroethylene resins, fluoroalkyl polysiloxanes, fluorinated graphite, etc.; alcohol additives such as higher alcohols, etc.; and metallic compound additives such as naphthenates (lead naphthenates), fatty acid salts (fatty acid lead), thiophosphates (zinc dialkyl dithiophosphate), thiocarbamates, organomolybdenum compounds, organic tin compounds, organogermanium compounds, boric acid esters, etc.

As the chlorine capturing agents, there can be mentioned compounds having glycidyl ether group, epoxy fatty acid monoesters, epoxidized fats and oils, compounds having epoxy chloalkyl group, etc. As the antioxidants, there can be included phenols (2,6-di-tert-butyl-p-cresol), aromatic amines (α -naphthylamine), etc. As the metal deactivators, there can be mentioned benzotriazole derivatives, etc. As the defoaming agents, silicone oil (dimethylpolysiloxane), polymethacrylates, etc. can be included. As the detergent dispersants, sulfonates, phenates, succinimides, etc. can be included. As the viscosity index improvers, polymethacrylates, polyisobutylene, ethylene-propylene copolymer, hydrogenated styrene-diene copolymer, etc. can be exemplified.

In what follows, the present invention will be described in more detail by referring to Examples, which however, are not intended to limit the invention thereto.

Examples 1 to 18 and Comparative Examples 1 to 4

The measurements of the wear resistance between aluminum material and steel material, stability and critical miscibility temperature at elevated temperature region were made using a variety of polyoxyalkylene glycol derivatives blended with various additives as sample oils as listed in Table 1 by way of the following testing procedures. The results are shown in Table 1.

(1) Wear resistance

The wear loss was measured using aluminum (A 4032) as a block and steel (SUJ-2) as a pin in Falex wear test, at a Flon 134a blow rate of 10 liter/hour under a load of 500 pounds for test hour of one hour.

(2) Stability

The stability was evaluated by means of shield tube test. A mixture of the sample oil and the refrigerant (Flon 134a) (2:1) was sealed up in a glass tube into which iron, copper and aluminum catalysts were inserted. After heating for 720 hours at 175 °C, the oil and catalysts were observed for appearance to check whether or not a sludge was formed.

(3) Critical miscibility temperature at elevated temperature region

A mixture of the sample oil and the refrigerant (Flon 134a) (1:9 by weight) was sealed up in a pressure glass vessel with an inner volume of about 10 ml. When the mixture was uniformly mingled with each other, its temperature was gradually raised. Thus the initial temperature of phase separation of the sample oil from the refrigerant was measured to determine the critical miscibility temperature at elevated temperature region.

Table 1

No.	Polyoxyalkylene glycol derivative	Composition				Species	compounding ratio (wt%)	(b) Component		Species	compounding ratio (wt%)
		(a) Component		compounding ratio (wt%)							
		Species	compounding ratio (wt%)	Species	compounding ratio (wt%)						
Example 1	polyoxypropylene glycol	sorbitan	1	monooleate		TCP*7	1				
Example 2	dimethyl ether*1	sorbitan	2	monooleate		TCP*7	1				
Example 3	polyoxypropylene glycol	sorbitan	5	monooleate		TCP*7	1				
Example 4	dimethyl ether*1	monooleate	10	monooleate		TCP*7	1				
Example 5	polyoxypropylene glycol	sorbitan	1.5	monooleate		TCP*7	1				
Example 6	dimethyl ether*1	glycerol	3	monooleate		TCP*7	1				
Example 7	polyoxypropylene glycol	glycerol	1	monooleate		TCP*7	1				
Example 8	dimethyl ether*2	sorbitan	1	monooleate		TCP*7	1				
Example 9	polyoxypropylene glycol	glycerol	1	monooleate		TCP*7	1				
Example 10	dimethyl ether*2	monooleate	1	monooleate		TCP*7	1				
Example 11	polyoxypropylene glycol	sorbitan	1	monooleate		TCP*7	1				
Example 12	monomethyl ether*3	monooleate	1	monooleate		TCP*7	1				
Example 13	polyoxypropylene glycol	glycerol	1	monooleate		TCP*7	1				
Example 14	monomethyl ether*3	sorbitan	1	monooleate		TCP*7	1				
	polyoxypropylene glycol	glycerol	1	monooleate		TCP*7	1				
	polyoxypropylene glycol	monooleate	1	monooleate		TCP*7	1				
	glycerol derivative	sorbitan	1	monooleate		TCP*7	1				
	trimethyl ether*5	monooleate	1	monooleate		TCP*7	1				
	polyoxypropylene glycol	sorbitan	1	monooleate		TCP*7	1				
	monobutyl ether*6	monooleate	1	monooleate		TCP*7	1				

Table 1 (continued 1)

No.	Polyoxyalkylene glycol derivative	Component			
		(a) Component		(b) Component	
		Species	Compounding ratio (wt%)	Species	Compounding ratio (wt%)
Example 15	polyoxypropylene glycol dimethyl ether*1	sorbitan monooleate	2	TOP*8	1
Example 16	polyoxypropylene glycol dimethyl ether*1	sorbitan monooleate	1	DOHP*9	1
Example 17	PO·EO(9:1) copolymer glycol dimethyl ether*10	sorbitan monooleate	1	TCP*7	1
Example 18	PO·EO(8:2) copolymer glycol dimethyl ether*11	sorbitan monooleate	1	TCP*7	1
Comparative Example 1	polyoxypropylene glycol dimethyl ether*1	sorbitan monooleate	5	-	-
Comparative Example 2	polyoxypropylene glycol dimethyl ether*1	sorbitan monooleate	-	TCP*7	1
Comparative Example 3	polyoxypropylene glycol dimethyl ether*1	dioctyl adipate	10	TCP*7	1
Comparative Example 4	polyoxypropylene glycol dimethyl ether*1	chlorinated paraffin	1.5	-	-

Table 1 (continued 2)

No.	Wear resistance (mg)	Performance		
		Appearance	Stability	Critical miscibility temperature at elevated temperature region (°C)
Example 1	1.0	good	good	60 or higher
Example 2	0.6	good	good	60 or higher
Example 3	0.2	good	good	60 or higher
Example 4	0.2	good	good	60 or higher
Example 5	20	good	good	60 or higher
Example 6	8	good	good	60 or higher
Example 7	1.5	good	good	60 or higher
Example 8	23	good	good	60 or higher
Example 9	2.0	good	good	60 or higher
Example 10	3.1	good	good	60 or higher
Example 11	2.5	good	good	60 or higher
Example 12	2.9	good	good	60 or higher
Example 13	1.8	good	good	60 or higher
Example 14	2.1	good	good	60 or higher
				56

Table 1 (continued 3)

No.	Wear resistance (mg)	Performance			Critical miscibility temperature at elevated temperature region (°C)
		Appearance	Stability Catalyst	Sludge formation	
Example 15	0.5	good	good	none	60 or higher
Example 16	0.4	good	good	none	60 or higher
Example 17	0.6	good	good	none	60 or higher
Example 18	0.6	good	good	none	60 or higher
Comparative Example 1	116	good	good	none	60 or higher
Comparative Example 2	65	good	good	none	60 or higher
Comparative Example 3	61	good	good	none	60 or higher
Comparative Example 4	27	brown	color change (corroded)	formed	60 or higher

*1 Average molecular weight 1270, *2 Average molecular weight 640,
 *3 Average molecular weight 1030, *4 Average molecular weight 1100,
 *5 Average molecular weight 1200, *6 Average molecular weight 1100,
 *7 Tricresyl phosphate, *8 Trioctyl phosphate, *9 Dioctyl hydrogen phosphite,
 *10 PO and EO indicate propylene oxide and ethylene oxide, respectively.
 This ether has an average molecular weight of 1300.
 *11 PO and EO indicate propylene oxide and ethylene oxide, respectively.
 This ether has an average molecular weight of 1300.

Examples 19 to 33 and Comparative Examples 5 to 8

The measurements of the wear resistance between aluminum material and steel material, stability and critical miscibility temperature at elevated temperature region were made using a variety of polyester compounds blended with various additives as sample oils as listed in Table 2 by way of the following testing procedures. The results are shown in Table 2.

(1) Wear resistance

The wear loss was measured using aluminum (A 4032) as a block and steel (SUJ-2) as a pin in Falex wear test, at a Flon 134a blow rate of 10 liter/hour under a load of 400 pounds for a test hour of one hour at

a revolution of 1200 rpm at an oil temperature of 80° C.

(2) Stability

- 6 The stability was evaluated by means of shield tube test. A mixture of the sample oil and the refrigerant (Flon 134a) (2:1) was sealed up in a glass tube into which iron, copper and aluminum catalysts were inserted. After heating for 720 hours at 175° C, the oil and catalysts were observed for appearance to check whether or not a sludge was formed.

10 (3) Critical miscibility temperature at elevated temperature region

- A mixture of the sample oil and the refrigerant (Flon 134a) (1:9 by weight) was sealed up in a pressure glass vessel with an inner volume of about 10 ml. When the mixture was uniformly mingled with each other, its temperature was gradually raised. Thus the initial temperature of phase separation of the sample oil from
15 the refrigerant was measured to determine the critical miscibility temperature at elevated temperature region.

(4) Saturated moisture

- 20 The sample oil and water each of 20 ml were mixed with stirring and were allowed to stand to separate the oil phase from the water phase. Thus the moisture in the oil phase was measured.

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Table 2

No.	Polyester compound	Composition			
		(a) Component		(b) Component	
		Species	compounding ratio (wt%)	Species	compounding ratio (wt%)
Example 19	ester compound I*1	sorbitan monooleate	1	TCP	1
Example 20	ester compound I*1	sorbitan monooleate	2	TCP	1
Example 21	ester compound I*1	sorbitan monooleate	5	TCP	1
Example 22	ester compound I*1	sorbitan monooleate	10	TCP	1
Example 23	ester compound I*1	sorbitan monooleate	1.5	TCP	1
Example 24	ester compound I*1	glycerol monooleate	3	TCP	1
Example 25	ester compound II*2	sorbitan monooleate	3	TCP	1
Example 26	ester compound II*2	glycerol monooleate	3	TCP	1
Example 27	ester compound III*3	sorbitan monooleate	3	TCP	1
Example 28	ester compound III*3	glycerol monooleate	3	TCP	1
Example 29	ester compound IV*4	sorbitan monooleate	3	TCP	1
Example 30	ester compound IV*4	glycerol monooleate	3	TCP	1

Table 2 (continued 1)

NO.	Wear resistance (mg)	Performance			Saturated moisture (wt%)
		Appearance	Stability Catalyst	Sludge formation	
Example 19	90	good	good	none	0.2
Example 20	82	good	good	none	0.2
Example 21	0.5	good	good	none	0.2
Example 22	0.2	good	good	none	0.2
Example 23	73	good	good	none	0.2
Example 24	22	good	good	none	0.2
Example 25	43	good	good	none	0.2
Example 26	50	good	good	none	0.2
Example 27	41	good	good	none	0.2
Example 28	49	good	good	none	0.2
Example 29	30	good	good	none	0.2
Example 30	38	good	good	none	0.2

Table 2 (continued 2)

No.	Polyester compound	Composition			
		(a) Component		(b) Component	
		Species	Compounding ratio (wt%)	Species	Compounding ratio (wt%)
Example 31	mixture of polyoxypropylene glycol dimethyl ether*5 and ester compound*3 (1:1)	sorbitan monooleate	5	TCP	1
Example 32	ester compound II*2	sorbitan monooleate	5	TOP	1
Example 33	ester compound I*1	sorbitan monooleate	5	DOHP	1
Comparative Example 5	ester compound I*1	sorbitan monooleate	5	-	-
Comparative Example 6	ester compound I*1	-	-	TCP	1
Comparative Example 7	ester compound I*1	dioctyl adipate	10	TCP	1
Comparative Example 8	ester compound I*1	chlorinated paraffin	1.0	-	-

Table 2 (continued 3)

No.	Wear resistance (mg)	Performance			Saturated moisture (wt%)
		Appearance	Stability Catalyst	Sludge formation	
Example 31	0.2	good	good	none	1.0
Example 32	0.2	good	good	none	0.2
Example 33	0.2	good	good	none	0.2
Comparative Example 5	Seizing	good	good	good	0.2
Comparative Example 6	190	good	good	good	0.2
Comparative Example 7	Seizing	good	good	good	0.2
Comparative Example 8	20	brown	color change	formed	0.2

- *1 ester compound I : polyester from neopentyl glycol, adipic acid and 2-methyl caprate
 kinematic viscosity 90.5 cst (40°C), pour point -50°C or lower
- *2 ester compound II : hexaester from dipentaerythritol and isovaleric acid
 kinematic viscosity 70.5 cst (40°C), pour point -45°C
- *3 ester compound III : trimethylol propane trisonanate (Product of Ciba-Geigy, trademark
 Reolube LR-2700
 kinematic viscosity 20.3 cst (40°C), pour point -50°C or lower
- *4 ester compound IV : polyester from poly(1,2-propylene)glycol, adipic acid and methyl
 alcohol
 kinematic viscosity 33.2 cst (40°C), pour point -50°C or lower
- *5 average molecular weight of polyoxypropylene glycol dimethyl ether : 1270

INDUSTRIAL APPLICABILITY

The refrigerator oil composition according to the present invention is excellent in stability, miscibility with a refrigerant of a hydrogen-containing hydrofluorocarbon such as Flon 134a and lubrication performance in the atmosphere of the above refrigerant, and functions quite effectively to improve wear resistance especially between aluminum material and steel material with an additional advantage of low hygroscopicity. Accordingly, the refrigerator oil composition according to the present invention is utilized as

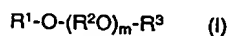
a lubricating oil for a various types of refrigerators using a hydrogenated Flon compound as a refrigerant as well as a compression type refrigerator. Above all, the refrigerator oil composition according to the present invention has satisfactory miscibility with hydrogenated Flon compounds (hydrogenated fluoroalkane) such as Flon 134a; 1,1,2,2-tetrafluoroethane (Flon 134); 1,1-dichloro-2,2,2-trifluoroethane (Flon 123); 1-chloro-1,1-difluoroethane (Flon 142b); 1-1-difluoro ethane (Flon 152a); chlorodifluoromethan (Flon 22), trifluoromethan (Flon 23), etc.

Accordingly, the refrigerator oil composition according to the present invention is expected to be effectively utilized as a lubricating oil for refrigerators using various types of hydrogenated Flon compounds as refrigerant, electric refrigerators or freezers, coolers (especially, air conditioner for car), heat pumps, etc.

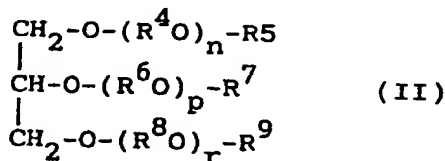
Claims

1. A refrigerator oil composition for hydrogen-containing hydrofluorocarbon refrigerant which comprises at least one compound selected from (A) a polyoxyalkylene glycol derivative, and (B) polyester compounds having a kinematic viscosity at 40 °C of 5 to 1000 cSt and at least two ester linkages which are compounded with (a) an aliphatic acid partially esterified with a polyhydric alcohol, and (b) at least one compound selected from among phosphate compounds and phosphite compounds.

2. The refrigerator oil composition according to Claim 1, wherein said (A) polyoxyalkylene glycol derivative is represented by the formula



and/or



wherein R^1 , R^3 , R^5 , R^7 and R^9 are each a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an alkenyl group having 2 to 20 carbon atoms, an alkynyl group having 2 to 20 carbon atoms, an acyl group having 1 to 20 carbon atoms, an aryl group having 6 to 20 carbon atoms, an aralkyl group having 7 to 20 carbon group or a cycloalkyl group having 4 to 20 carbon atoms; R^2 , R^4 , R^6 and R^8 are each an alkylene group having 2 to 4 carbon atoms; m is an integer from 3 through 100; and n , p and r are each a positive integer simultaneously satisfying the relationship

$$3 \leq n + p + r \leq 100.$$

3. The refrigerator oil composition according to Claim 1, wherein said (B) polyester compound is a reaction product of (i) a polybasic carboxylic acid or derivative thereof; (ii) a polyhydric alcohol or derivative thereof; and (iii) a monobasic aliphatic acid or derivative thereof.
4. The refrigerator oil composition according to Claim 1, wherein said (B) polyester compound is a reaction product of (i) a polybasic carboxylic acid or derivative thereof; (ii) a polyhydric alcohol or derivative thereof; and (iv) a monohydric aliphatic alcohol or derivative thereof.
5. The refrigerator oil composition according to Claim 1, wherein said (B) polyester compound is a reaction product of (ii) a polyhydric alcohol or derivative thereof; and (iii) a monobasic aliphatic acid or derivative thereof.
6. The refrigerator oil composition according to Claim 1, wherein said (B) polyester compound is a reaction product of (iv) a monohydric aliphatic alcohol or derivative thereof; and (i) a polybasic carboxylic acid or derivative thereof.

7. The refrigerator oil composition according to Claim 1, where said composition contains 0.1 to 10% by weight of said (a) an aliphatic acid partially esterified with a polyhydric alcohol and 0.1 to 5% by weight of said (b) at least one compound selected from among phosphate compounds and phosphite compounds.
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8. The refrigerator oil composition according to Claim 3 or 5, wherein said (iii) a monobasic aliphatic acid or derivative thereof is that which contains a branched-chain alkyl group of 3 to 18 carbon atoms.
9. The refrigerator oil composition according to Claim 4 or 6, wherein said (iv) a monohydric aliphatic alcohol or derivative thereof is that which contains a branched-chain alkyl group having 3 to 18 carbon atoms.
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10. A method for effecting lubrication in a compression-type refrigerator using a hydrogen-containing hydrofluorocarbon as a refrigerant characterized in that the lubrication is effected by the use of said refrigerator oil composition as defined in Claim 1.
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11. A compression-type refrigeration system which comprises a compressor, hydrogen-containing hydrofluorocarbon as a refrigerant and said refrigerator oil composition as defined in Claim 1.
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INTERNATIONAL SEARCH REPORT

International Application No PCT/JP90/01469

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) * According to International Patent Classification (IPC) or to both National Classification and IPC Int. Cl. ⁵ C10M169/04// (C10M169/04, C10M107:34, C10M129:76, C10M137:04), C10N40:30		
II. FIELDS SEARCHED Minimum Documentation Searched ? Classification System : Classification Symbols IPC C10M129/76, C10M137/04-10, C10M107/34, C10M105/18, C10N40:30 Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched *		
III. DOCUMENTS CONSIDERED TO BE RELEVANT *		
Category *	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³
Y	EP, A, 336,171 (Nippon Oil Co., Ltd.), October 11, 1989 (11. 10. 89), Claims & JP, A, 2-43290 & JP, A, 2-84491	1-11
Y	JP, A, 1-271491 (Nippon Oil Co., Ltd.), October 30, 1989 (30. 10. 89), Claim (Family: none)	1-11
A	EP, A, 17,072 (DOW CHEMICAL Co.), October 15, 1980 (15. 10. 80) & US, A, 4,302,343 & JP, A, 55-133489	1-11
A	JP, A, 56-26997 (Showa Aluminum Corp.), March 16, 1981 (16. 03. 81), (Family: none)	1-11
A	JP, A, 61-60791 (Hitachi, Ltd.), March 28, 1986 (28. 03. 86), (Family: none)	1-11
* Special categories of cited documents: ¹⁴ "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step "Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search January 28, 1991 (28. 01. 91)		Date of Mailing of this International Search Report February 12, 1991 (12. 02. 91)
International Searching Authority Japanese Patent Office		Signature of Authorized Officer